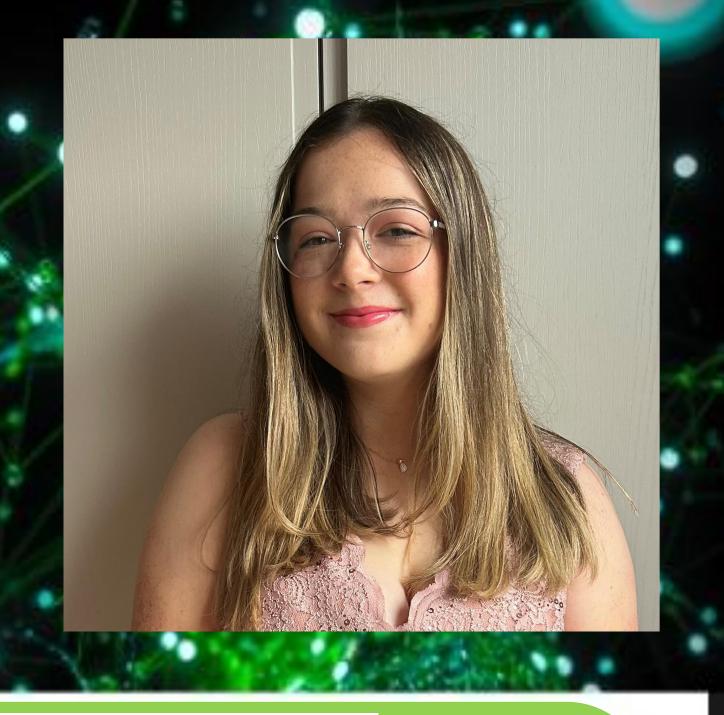
QUEEN'S UNIVERSITY IONIC LIQUID LABORATORIES

QUILL

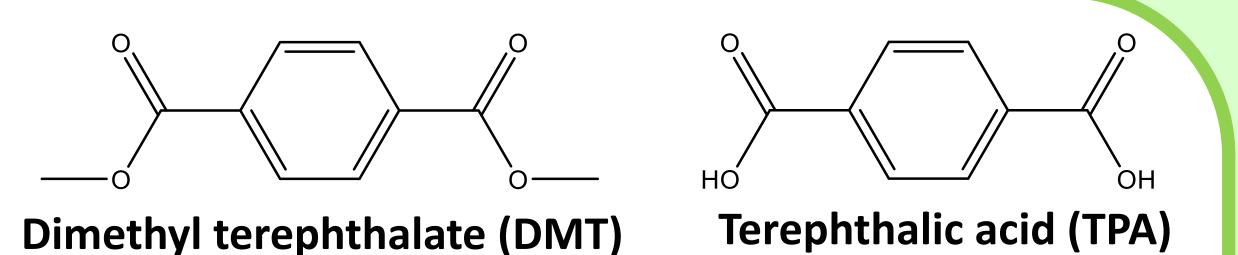


Methanolysis of PET using Ionic Liquid Catalyst

Abbi Armstrong*, Emma McCrea, Prof. Małgorzata Swadźba-Kwaśny, Prof. John Holbrey The QUILL Research Centre, David Keir Building, Stranmillis Road, Belfast, UK. *aarmstrong32@qub.ac.uk

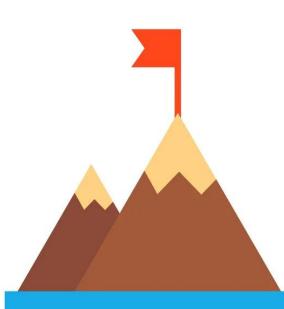
1. Introduction

- PET can be chemically recycled into its monomers, dimers or oligomers. This can be done by a range of processes, our focus is methanolysis to produce DMT (dimethyl terephthalate), hydrolysis to produce TPA (terephthalic acid), and glycolysis to produce BHET (bis(2-hydroxyethyl)-terephthalate).
- The current mechanical recycling process results in weaker plastics which end up being unfit for use. Chemical recycling aims to restore plastic back to its monomer, so plastic use can be a closed loop system.



Bis(2-Hydroxyethyl) terephthalate (BHET)

2. Challenges and goals



Energy intensive

Burning in reactor **Isolation of** products

Purity of products

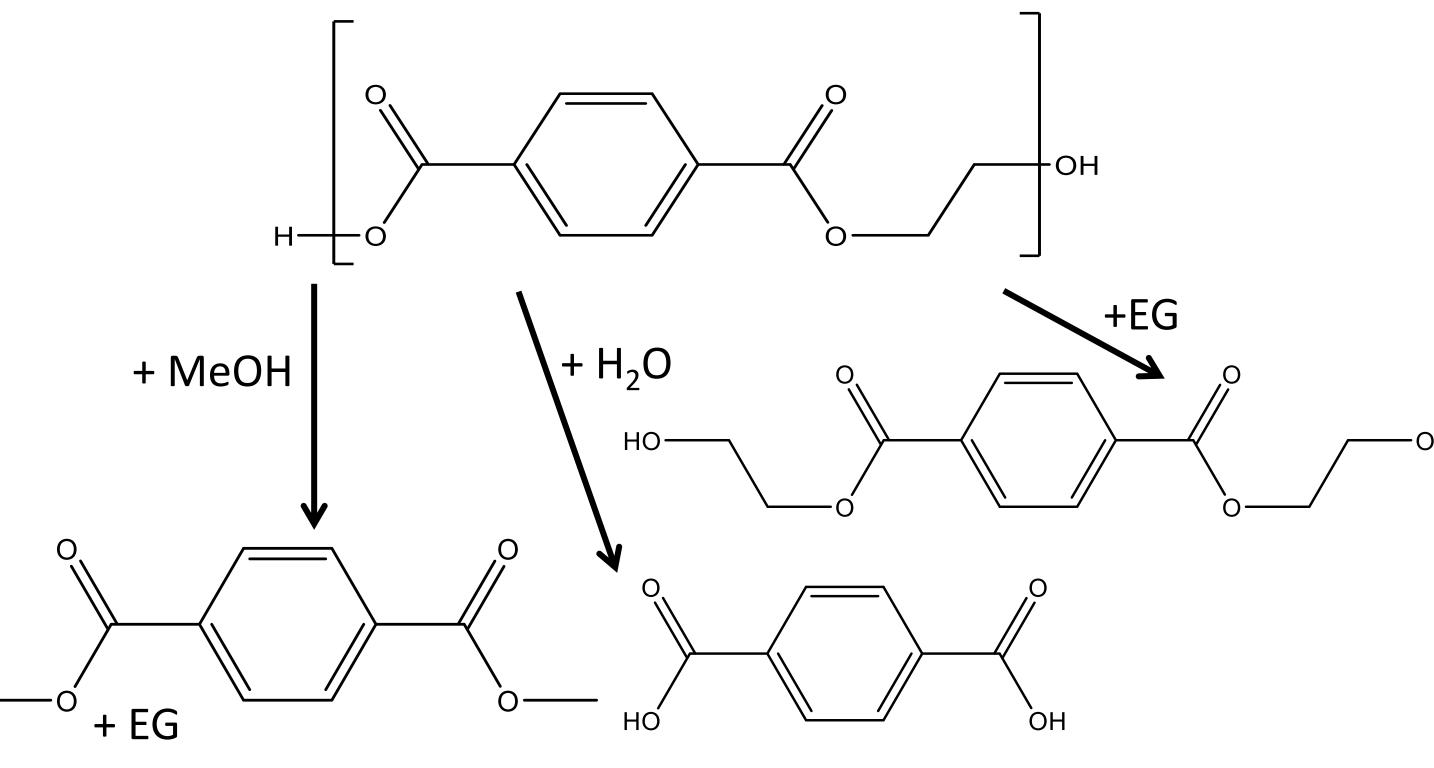


To determine the most efficient way to preform methanolysis, hydrolysis and glycolysis using [HN₂₂₂][HSO₄].H₂SO₄ and to determine the best way to extract the product.

Benefits:

Robust, cheap IL catalyst Relationship between ILs and microwaves for efficient heating¹

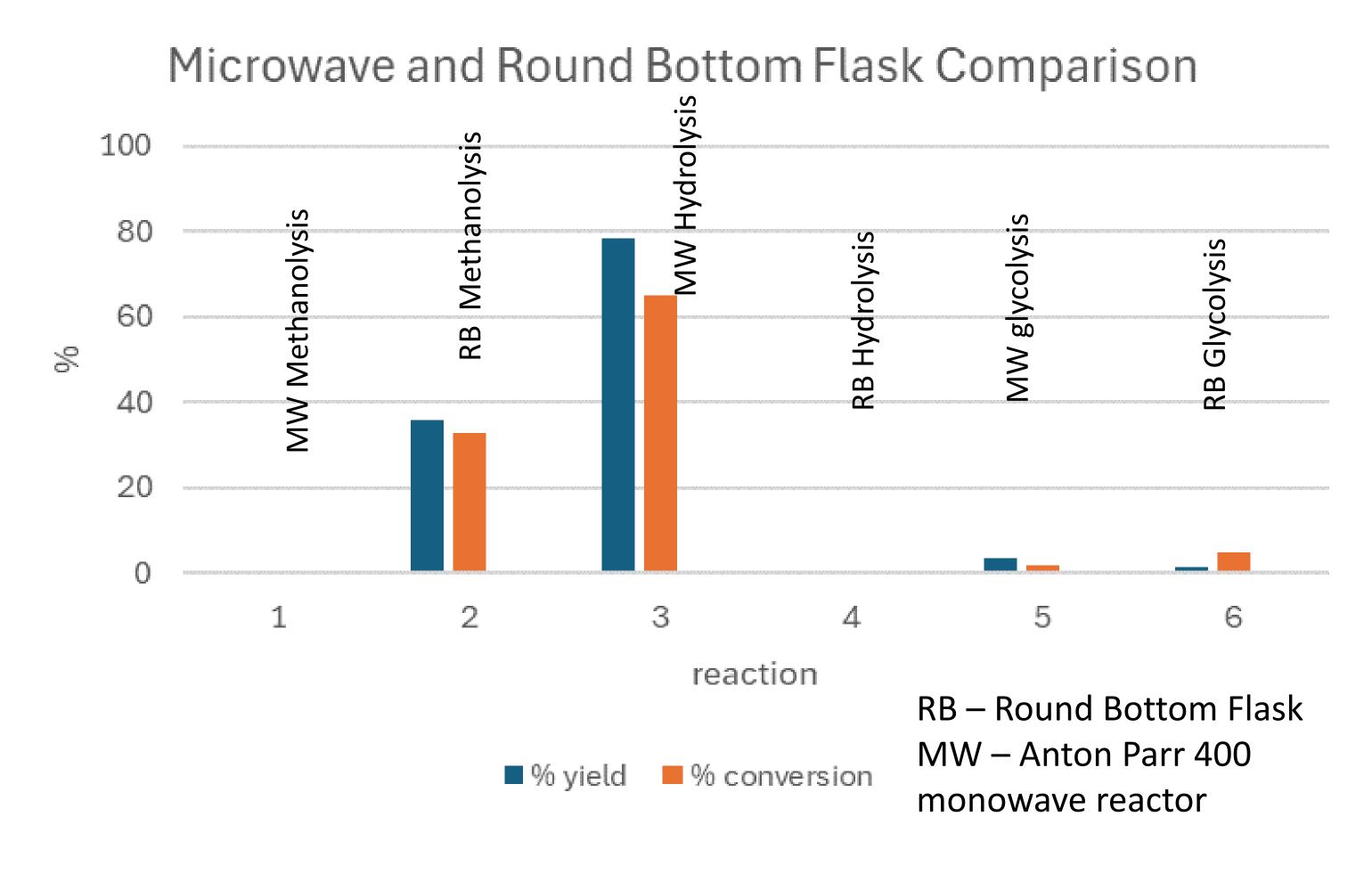
3. Reactions

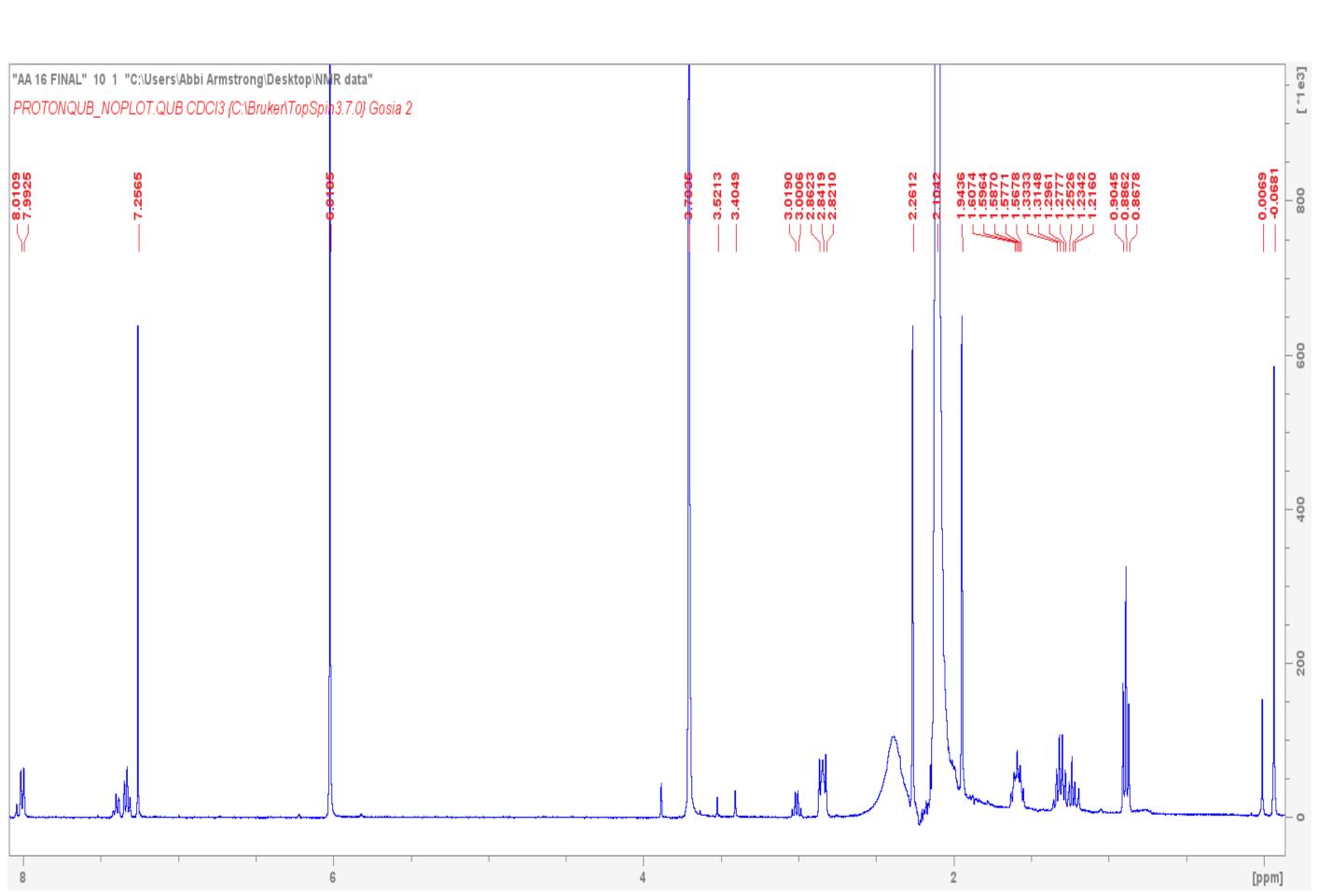


[HN₂₂₂][HSO₄].H₂SO₄ Anton Parr 400 monowave reactor EG – Ethylene glycol

4. Results

1g PET, 3g [HN₂₂₂][HSO₄].H₂SO₄. 5g H₂O / EG / MeOH, 3h 180°C, (round bottom glycolysis conducted at 120° C due to burning)





¹H NMR (400 MHz) CDCl₃

5. Conclusion and References

- The IL produces TPA at low catalyst loading using the Anton parr monowave 400 reactor
- The IL produces DMT at low catalyst loading when heated in a round bottom flask equipped with a condenser
- Successful analysis of product by NMR
- Future work could include investigating colours produced and further glycolysis methods
- 1. Chem. Rev., 2024, 124, 2651–2698, Q. F. Yue, C. X. Wang, L. N. Zhang, Y. Ni and 2. Y. X. Jin, Polym Degrad Stab, 2011, 96, 399–403

Many Thanks to QUILL Summer Studentship for funding my project